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INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<p>(51) International Patent Classification ⁵ : C07D 401/12</p>	<p>A1</p>	<p>(11) International Publication Number: WO 91/18895</p> <p>(43) International Publication Date: 12 December 1991 (12.12.91)</p>		
<table style="width: 100%; border: none;"> <tr> <td style="width: 50%; border: none; vertical-align: top;"> <p>(21) International Application Number: PCT/SE91/00402</p> <p>(22) International Filing Date: 5 June 1991 (05.06.91)</p> <p>(30) Priority data: 9002043-9 7 June 1990 (07.06.90) SE</p> <p>(71) Applicant: AKTIEBOLAGET ASTRA [SE/SE]; S-151 85 Södertälje (SE).</p> <p>(72) Inventor: BRÄNDSTRÖM, Arne, Elof ; Anders Mattssonsgatan 13 B, S-415 06 Göteborg (SE).</p> <p>(74) Agents: DANIELSSON, Sten et al.; AB Astra, Patent Department, S-151 85 Södertälje (SE).</p> </td> <td style="width: 50%; border: none; vertical-align: top;"> <p>(81) Designated States: AT, AT (European patent), AU, BB, BE (European patent), BF (OAPI patent), BG, BJ (OAPI patent), BR, CA, CF (OAPI patent), CG (OAPI patent), CH, CH (European patent), CI (OAPI patent), CM (OAPI patent), DE, DE (European patent), DK, DK (European patent), ES, ES (European patent), FI, FR (European patent), GA (OAPI patent), GB, GB (European patent), GN (OAPI patent), GR (European patent), HU, IT (European patent), JP, KP, KR, LK, LU, LU (European patent), MC, MG, ML (OAPI patent), MR (OAPI patent), MW, NL, NL (European patent), NO, PL, RO, SD, SE, SE (European patent), SN (OAPI patent), SU, TD (OAPI patent), TG (OAPI patent).</p> <p>Published <i>With international search report.</i></p> </td> </tr> </table>			<p>(21) International Application Number: PCT/SE91/00402</p> <p>(22) International Filing Date: 5 June 1991 (05.06.91)</p> <p>(30) Priority data: 9002043-9 7 June 1990 (07.06.90) SE</p> <p>(71) Applicant: AKTIEBOLAGET ASTRA [SE/SE]; S-151 85 Södertälje (SE).</p> <p>(72) Inventor: BRÄNDSTRÖM, Arne, Elof ; Anders Mattssonsgatan 13 B, S-415 06 Göteborg (SE).</p> <p>(74) Agents: DANIELSSON, Sten et al.; AB Astra, Patent Department, S-151 85 Södertälje (SE).</p>	<p>(81) Designated States: AT, AT (European patent), AU, BB, BE (European patent), BF (OAPI patent), BG, BJ (OAPI patent), BR, CA, CF (OAPI patent), CG (OAPI patent), CH, CH (European patent), CI (OAPI patent), CM (OAPI patent), DE, DE (European patent), DK, DK (European patent), ES, ES (European patent), FI, FR (European patent), GA (OAPI patent), GB, GB (European patent), GN (OAPI patent), GR (European patent), HU, IT (European patent), JP, KP, KR, LK, LU, LU (European patent), MC, MG, ML (OAPI patent), MR (OAPI patent), MW, NL, NL (European patent), NO, PL, RO, SD, SE, SE (European patent), SN (OAPI patent), SU, TD (OAPI patent), TG (OAPI patent).</p> <p>Published <i>With international search report.</i></p>
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<p>(54) Title: IMPROVED METHOD FOR SYNTHESIS</p> <p>(57) Abstract</p> <p>The present invention relates to an improved method for the synthesis of omeprazole, comprising the steps of reacting 5-methoxy-2-[(4-methoxy-3,5-dimethyl-2-pyridinyl)-methylthio]-1H-benzimidazole with m-chloroperoxy-benzoic acid in a methylene chloride solution at a substantially constant pH of about 8.0 to 8.6; extracting the reaction mixture with aqueous NaOH; separating the aqueous phase from the organic phase; and adding an alkyl formate to the aqueous phase, resulting in crystallization of omeprazole.</p>				

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Improved method for synthesisTechnical field

- 5 The present invention relates to an improved method for the synthesis of 5-methoxy-2-[[[4-methoxy-3,5-dimethyl-2-pyridinyl)-methyl]sulfinyl-1H-benzimidazole, referred to under its generic name omeprazole throughout the following specification and claims.

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Prior art

- US-A-4 255 431 discloses a process for the synthesis of omeprazole comprising the steps of reacting 5-methoxy-2-
15 [(4-methoxy-3,5-dimethyl-2-pyridinyl)-methylthio]-1H-benzimidazole in a methylene chloride solution with m-chloroperoxybenzoic acid resulting in the formation of omeprazole and m-chlorobenzoic acid. omeprazole is highly sensitive to acids, and the reaction mixture has to be
20 maintained at a low temperature to prevent excessive decomposition in the reaction mixture.

- The product is worked-up by filtering-off of m-chlorobenzoic acid formed during the reaction. The filtrate is
25 diluted with methylene chloride, is extracted with Na_2CO_3 solution, dried and evaporated. The resulting omeprazole product is contaminated with starting materials and by-products.

30 Summary of the invention

- The object of the present invention is to provide an improved method for the synthesis of omeprazole, which eliminates the drawbacks of previously known methods.

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This object is achieved according to the present inven-

tion, which is characterized by the steps of reacting 5-methoxy-2-[(4-methoxy-3,5-dimethyl-2-pyridinyl)-methylthio]-1H-benzimidazole (below denoted Compound I) with m-chloroperoxybenzoic acid in a methylene chloride solution
5 at a substantially constant pH of about 8.0 to 8.6; extracting the reaction with aqueous NaOH; separating the aqueous phase from the organic phase; and adding an alkyl formate to the aqueous phase, resulting in crystallization of omeprazole.

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The m-chloroperoxybenzoic acid is suitably used in an amount of 0.7 - 1.4 molar equivalents of Compound I, and preferably in an amount of 0.9 - 1.2 molar equivalents.

15 According to one embodiment of the invention, the alkyl formate is methylformate or ethylformate, methylformate being preferred.

The alkyl formate is suitably used in an amount of 1.2 -
20 2.0 molar equivalents of Compound I, and preferably in an amount of 1.5 - 1.8 molar equivalents.

One important feature of the method according to the invention is that unreacted sulfide is not transferred
25 into the aqueous phase upon the extraction with aqueous NaOH. Another important feature is that m-chlorobenzoic acid does not crystallize upon the addition of methylformate to the aqueous phase, thereby eliminating the need of filtering-off of m-chlorobenzoic acid in a previous
30 step.

The pH of the reaction mixture may be maintained within the pH range of 8.0 - 8.6 with the aid of pH static titration with NaOH or with the use of a buffer. Preferred
35 buffers are sodium bicarbonate and potassium bicarbonate.

A great advantage of the method according to the invention is that the reaction takes place in the organic methylene chloride phase while the m-chlorobenzoic acid formed during the reaction goes into the aqueous phase containing the buffer, in the case a buffer is used. Because of this, omeprazole formed does not stay in contact with the acid and the reaction may be performed at a temperature above 0°C.

10 According to one embodiment of the invention the pH of the aqueous NaOH phase is kept at above about 12.

According to another embodiment of the invention the crystallization of omeprazole is performed at a pH of above 9.

The invention will be further illustrated below with a non-limiting example.

20 Example

5-methoxy-2-[(4-methoxy-3,5-dimethyl-2-pyridinyl)-methylthio]-1H-benzimidazole (16.2 g; 0.0492 mol) is reacted with m-chloroperoxybenzoic acid (13.6 g; 0.0537 mol) in CH₂Cl₂ acting as a solvent at a pH of 8.6, which is maintained by the presence of KHCO₃ (5.6 g; 0.056 mol) acting as a buffer. The temperature is maintained at about 0°C during the addition.

30 Diluted NaOH is added to a pH above 12 and the CH₂Cl₂ phase is separated off.

Methylformate (4.7 g) is charged to the water phase and the pH is kept above 9, whereupon omeprazole crystallizes. The crystals are filtered off and are washed with water and methanol at a temperature of about 0°C. The washed crystals are dried under vacuum. Yield: 15.6 g (92 %).

C l a i m s

1. An improved method for the synthesis of omeprazole, characterized by the steps of reacting 5-methoxy-2-[(4-methoxy-3,5-dimethyl-2-pyridinyl)-methylthio]-1H-benzimidazole (Compound I) with m-chloroperoxybenzoic acid in a methylene chloride solution at a substantially constant pH of about 8.0 to 8.6; extracting the reaction mixture with aqueous NaOH; separating the aqueous phase from the organic phase; and adding an alkyl formate to the aqueous phase, resulting in the crystallization of omeprazole.
2. Method according to claim 1, characterized in that the m-chloroperoxybenzoic acid is used in an amount of 0.7 - 1.4, preferably 0.9 - 1.2, molar equivalents of Compound I.
3. Method according to claim 1 or 2, characterized in that the alkyl formate is methylformate.
4. Method according to claims 1 - 3, characterized in that pH of the reaction mixture is maintained within the pH range of 8.0 - 8.6 with the aid of pH static titration with NaOH.
5. Method according to claims 1 - 4, characterized in that pH of the reaction mixture is maintained within the pH range of 8.0 - 8.6 with the use of a buffer.
6. Method according to claim 5, characterized in that the buffer is sodium bicarbonate or potassium bicarbonate.
7. Method according to claims 1 - 6, characterized in that the pH of the aqueous NaOH phase is

kept at above about 12.

8. Method according to claims 1 - 7, c h a r a c t e -
r i z e d in that the alkyl formate is added in an amount
5 of 1.2 - 2.0, preferably 1.5 - 1.8, molar equivalents of
Compound I.

9. Method according to claims 1 - 8, c h a r a c t e -
r i z e d in that the crystallization of omeprazole is
10 performed at a pH of above 9.

INTERNATIONAL SEARCH REPORT

International Application No PCT/SE 91/00402

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) ⁶ According to International Patent Classification (IPC) or to both National Classification and IPC IPC5: C 07 D 401/12						
II. FIELDS SEARCHED <div style="text-align: right; margin-right: 50px;">Minimum Documentation Searched⁷</div> <table style="width: 100%; border: none;"> <tr> <td style="width: 20%; border: none;">Classification System</td> <td style="border: none;">Classification Symbols</td> </tr> <tr> <td style="border: 1px solid black; padding: 5px;">IPC5</td> <td style="border: 1px solid black; padding: 5px;">C 07 D</td> </tr> </table>			Classification System	Classification Symbols	IPC5	C 07 D
Classification System	Classification Symbols					
IPC5	C 07 D					
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in Fields Searched ⁸ SE,DK,FI,NO classes as above						
III. DOCUMENTS CONSIDERED TO BE RELEVANT⁹						
Category *	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³				
A	US, A, 4255431 (AKTIEBOLAGET HÄSSLE) 10 March 1981, see example 1 --	1				
A	US, A, 4182766 (HOFFMANN-LA-ROCHE INC) 8 January 1980, see example 17 --	1				
A	WO, A1, 8705021 (AKTIEBOLAGET HÄSSLE) 27 August 1987, see example 1 --	1				
A	EP, A, 0197013 (AKTIEBOLAGET HÄSSLE) 8 October 1986, see example 2 --	1				
<div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <p>* Special categories of cited documents:¹⁰</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the International filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the International filing date but later than the priority date claimed</p> </div> <div style="width: 45%;"> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance, the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance, the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&" document member of the same patent family</p> </div> </div>						
IV. CERTIFICATION						
Date of the Actual Completion of the International Search 28th August 1991	Date of Mailing of this International Search Report 1991 -09- 10					
International Searching Authority <div style="text-align: center;">SWEDISH PATENT OFFICE</div>	Signature of Authorized Officer Göran Karlsson					

III. DOCUMENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)		
Category -	Citation of Document, with indication, where appropriate, of the relevant passages	Relevant to Claim No
A	EP, A, 0163842 (F. HOFFMANN-LA ROCHE & CO.) 11 December 1985, see example 4 -- -----	1

**ANNEX TO THE INTERNATIONAL SEARCH REPORT
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Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A- 0163842	85-12-11	SU-A- 1396965	88-05-15
		US-A- 4634710	87-01-06
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